



Video Solution on Website:-

<https://physicsaholics.com/home/courseDetails/88>

Video Solution on YouTube:-

<https://youtu.be/yMof5Q3lttU>

Written Solution on Website:-

<https://physicsaholics.com/note/notesDetails/28>

- Q 1. A particle of mass $3m$ at rest decays into two particles of masses m and $2m$ having non-zero velocities. The ratio of the de-Broglie wavelengths of the particles $\left(\frac{\lambda_1}{\lambda_2}\right)$ is:
(a) $1/2$ (b) $1/4$ (c) 2 (d) None
- Q 2. The energy of a photon is equal to the kinetic energy of a proton. The energy of the photon is E . Let λ_1 be the de-Broglie wavelength of the proton and λ_2 be the wavelength of the photon. The ratio $\frac{\lambda_1}{\lambda_2}$ is proportional to:
(a) E^0 (b) $E^{1/2}$ (c) E^{-1} (d) E^{-2}
- Q 3. A beam of electron is used in an YDSE experiment. The slit width is d . When the velocity of electron is increased, then
(a) no interference is observed (b) fringe width increases
(c) fringe width decreases (d) fringe width remains same
- Q 4. If λ_p and λ_e denote the de-Broglie wavelength of proton and electron after they are accelerated from rest through the same potential difference, then –
(a) $\lambda_e = \lambda_p$ (b) $\lambda_e < \lambda_p$ (c) $\lambda_e > \lambda_p$ (d) $\lambda_e = \lambda_p/2$
- Q 5. The de Broglie wavelength of a bus moving with speed v is λ . Some passengers left the bus at a stoppage. Now when the bus moves with twice its initial speed. Now kinetic energy is found to be twice its initial value. What will be the de Broglie wavelength, now-
(a) λ (b) 2λ (c) $\lambda/2$ (d) $\lambda/4$
- Q 6. A monochromatic radiation of wavelength λ_1 is incident on a stationary atom as a result of which the wavelength of the photon after the collision becomes λ_2 and the recoiled atom has De Broglie's wavelength λ_3 . Then,
(a) $\lambda_3 = \sqrt{\lambda_1 \lambda_2}$
(b) $\lambda_1 = \frac{\lambda_2 \lambda_3}{\lambda_2 + \lambda_3}$
(c) $\lambda_1 = \sqrt{\lambda_1^2 + \lambda_2^2}$
(d) $\lambda_3 = \sqrt{\lambda_1^2 + \lambda_2^2}$
- Q 7. If E_1 , E_2 and E_3 represent respectively the kinetic energies of an electron, an alpha particle and a proton each having same de Broglie wavelength then:
(a) $E_1 > E_3 > E_2$ (b) $E_2 > E_3 > E_1$



(c) $E_1 > E_2 > E_3$

(d) $E_1 = E_2 = E_3$

Q 8. An electron of mass m , when accelerated through a potential difference V has de Broglie wavelength λ . The de Broglie wavelength associated with a proton of mass M when accelerated by same potential difference is

(a) $\lambda \sqrt{\frac{M}{m}}$

(b) $\lambda \sqrt{\frac{m}{M}}$

(c) $\lambda \frac{M}{m}$

(d) $\lambda \frac{m}{M}$

Q 9. A particle is moving in a closed orbit near origin, due to a force which is directed towards origin. The de Broglie wavelength of particle varies from λ_1 to λ_2 cyclically ($\lambda_1 > \lambda_2$). Then

(a) Particle could be moving in a circular orbit with centre at origin.

(b) Particle could be moving in an elliptical orbit with one focus at origin.

(c) When de Broglie wavelength is λ_1 , the particle is nearer to origin than when its value is λ_2 .

(d) When de Broglie wavelength is λ_2 , the particle is nearer to origin than when its value is λ_1

Q 10. The ratio of de Broglie wavelengths of proton and an alpha particle will be 1:2, if their

(a) kinetic energies are in ratio 1:8

(b) kinetic energies are in ratio 8:1

(c) Speeds are in ratio 1:8

(d) Speeds are in ratio 8:1

Answer Key

Q.1 d	Q.2 b	Q.3 c	Q.4 c	Q.5 a
Q.6 b	Q.7 a	Q.8 b	Q.9 b, d	Q.10 d